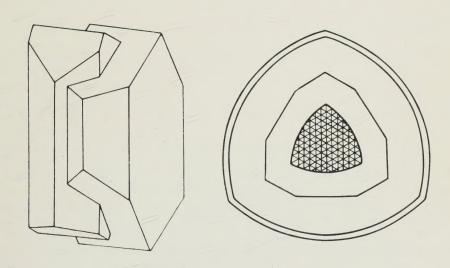
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# MINERALOGICAL ABSTRACTS

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## MINERALOGICAL ABSTRACTS

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## **ERRATA**

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92M/0877 92M/1195 text, line 1: for Alluair read Alluaiv citation: for J. J. Kisch read H. J. Kisch

92M/3437 92M/3853 text, line 1: for Kakum read Kakun text, lines 8: for cumengéite read cumengite

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## ORGANIZATION OF ABSTRACTS

Arising from a decision taken at the meeting of the International Mineralogical Association in Copenhagen in 1961 the Mineralogical Societies of America and Great Britain agreed to issue a joint statement to National Societies adhering to the Association inviting each Society to organize contributions of abstracts of papers published in the journals of its country on subjects relevant to *Mineralogical Abstracts*. This invitation was issued and has brought a gratifying response. Members of Societies which have agreed to co-operate in this way are entitled to receive *Mineralogical Abstracts* for their personal use at a reduced rate of subscription on application, which must be made through their National Society. The countries now co-operating include: Australia, Austria, Belgium, Bulgaria, Canada, Czechoslovakia, Denmark, Finland, France, Germany, India, Israel, Italy, Japan, Netherlands, New Zealand, Norway, Pakistan, Portugal, South Africa, Spain, Sweden, Switzerland, Turkey.

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# ABBREVIATIONS AND SYMBOLS

used in the text of abstracts

Andrew Land and the Control of the C				OPERCAL					
EMICAL & PHYSICAL CHE	MICA	L		OPTICAL					
tomic absorption spectrophotome	try		AAS	dispersion, e.g.					r > v
efore present			B.P.	extinction angle, e.g.					Y.C
ation-exchange capacity			c.e.c.	infrared				W. DOT	IR
oncentrated			conc.	optic axial angle					2V
ifferential thermal analysis			DTA	—, plane					O.A.P.
			dil.	refractive index in te					refr. ind.
ilute	***		EDA	— of isotropic min					n
nergy dispersive analysis	***	***			Clais		•••		"
lectron probe microanalysis	• • •		EPMA	refractive indices	,				
thylenediaminetetra-acetic acid	• • •	***	EDTA	of uniaxial miner				***	ω,ε
igacity			f	of biaxial minera		• • •			α,β,γ
as chromatography			GC	scanning electron mi				* * . *	SEM
eat of formation (absolute temper	ature su	bscript)	$\Delta Hf$	transmission electron	n micros	сору			TEM
ydrogen ion conc. acidity			pH	sign of biaxiality					
ductively coupled plasma	1		ICP	negative				07	2V <sub>a</sub> or -
nitial ratio, e.g				positive				9	2V <sub>y</sub> or +
		• • •	is all res	ultraviolet					UV
soluble residue		***	insol. res.	untaviolet	•••				OV
otopes, e.g	***		<sup>40</sup> Ar, <sup>40</sup> K	CHD. Dr. Stranger					
otope ratio, e.g		• • •	$\delta^{18}O$	PHYSICAL					
otope ratio normalised to chondr	ite, e.g.		$\varepsilon_{Nd}$						
rge ion lithophile	***		LIL	calculated					calc.
oss on ignition			ign. loss	cathodoluminescence	e			***	CL
nid-ocean ridge basalt			MORB	cycles per second					c/s
illiequivalent			me.	degree centigrade					°C
ass absorption spectrometry			MAS	degrees absolute			111		K
ass spectrometry			MS	density					D (quote units)
nicrogram				electron paramagneti					'1
			μg						e.p.r.
illion years		***	m.y.	hardness			• • •	•••	H.
eutron activation analysis			NAA						kbar
ot determined			n.d.						m.p.
ot found			nt. fd.	micron (10 <sup>-3</sup> mm)					μm
uclear magnetic resonance			NMR	nanometre (10 <sup>-6</sup> mm	)				nm
arts per billion (10 <sup>-9</sup> )			ppb	natural remanent ma					n.r.m.
arts per million			ppm						Ω
arts per trillion $(10^{-12})$			ppt	pressure(s)					P
atinum group elements, minerals			PGE, M	*				•••	7 .
		• • •							sol.
re earth elements		***	REE	specific gravity, term				own	sp.gr.
andard mean ocean water	***	4.4"	SMOW	4				***	T
rength of solution, normal			N	thermoluminescence					TL
—, molar			M	Vickers hardness nur	mber				VHN
ibstances in ionic state				wavelength					λ
anions, e.g			Cl <sup>-</sup> , SO <sub>4</sub> <sup>2</sup> -						
cations, e.g			K <sup>+</sup> , Fe <sup>3+</sup>						
ermogravimetric analysis			TGA	SYMBOLS					
ousand years									
		• • •	k.y.	approximately equal		• • •	***	* * *	~
		***	tr.	equal to		• • •			=
-ray powder diffraction	* * *	***	XRD	equal to or greater th	an			* * *	>
-ray fluorescence analysis	***	***	XRF	equal to or less than					<
				equilibrium					<b>=</b>
STALLOGRAPHIC & STR	UCTU	RAL		greater, less than					>, <
	00101		0	much greater, less th					
ngstrom unit (10 <sup>-8</sup> cm)			Å	41.1.		• • •	***	• • •	>>, <<
ystal axes		***	a, b, c			• • •	• • •		
— face indices			(hkl)	per cent	***	• • •	***	• • •	%
— form indices			{hkl}	per mille	***				%0
— zone indices			[hkl]	perpendicular to					1
dices of X-ray diffractions			hkl	proportional to					oc
		***							
tensity	***	***	1	GEOGRAPHICAL					
— relative	4.4.4		$I/I_{\rm o}$	GEOGRAFIICAL					
terplanar spacing			d	East Pacific rise					EPR
ica structural polymorphs			1 M <sub>1</sub> , 2 M	mid-Atlantic ridge					
nit cell, formula units			Z	east, eastern	***	***		***	MAR
——, repeat distances			a,b,c		• • • •			***	E
		• • •		north, northern	***				N
, reciprocal lattice length of	or edges		$a^*,b^*,c^*$	south, southern				***	S
——, interaxial angles direct la			α,β,γ						

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